

# Periodic Table of the Elements

<b>1</b> <b>H</b> 1.00794*															<b>1</b> <b>H</b> 1.00794*	<b>2</b> <b>He</b> 4.002602																													
<b>3</b> <b>Li</b> 6.941*	<b>4</b> <b>Be</b> 9.012182											<b>5</b> <b>B</b> 10.811*	<b>6</b> <b>C</b> 12.0107*	<b>7</b> <b>N</b> 14.00674*	<b>8</b> <b>O</b> 15.9994*	<b>9</b> <b>F</b> 18.9984032	<b>10</b> <b>Ne</b> 20.1797																												
<b>11</b> <b>Na</b> 22.98976928	<b>12</b> <b>Mg</b> 24.3050											<b>13</b> <b>Al</b> 26.9815386	<b>14</b> <b>Si</b> 28.0855*	<b>15</b> <b>P</b> 30.973762	<b>16</b> <b>S</b> 32.066*	<b>17</b> <b>Cl</b> 35.4527*	<b>18</b> <b>Ar</b> 39.948																												
<b>19</b> <b>K</b> 39.0983	<b>20</b> <b>Ca</b> 40.078	<b>21</b> <b>Sc</b> 44.955912	<b>22</b> <b>Ti</b> 47.867	<b>23</b> <b>V</b> 50.9415	<b>24</b> <b>Cr</b> 51.9961	<b>25</b> <b>Mn</b> 54.938045	<b>26</b> <b>Fe</b> 55.845	<b>27</b> <b>Co</b> 58.933195	<b>28</b> <b>Ni</b> 58.6934	<b>29</b> <b>Cu</b> 63.546	<b>30</b> <b>Zn</b> 65.38	<b>31</b> <b>Ga</b> 69.723	<b>32</b> <b>Ge</b> 72.63	<b>33</b> <b>As</b> 74.92160	<b>34</b> <b>Se</b> 78.96	<b>35</b> <b>Br</b> 79.904	<b>36</b> <b>Kr</b> 83.798																												
<b>37</b> <b>Rb</b> 85.4678	<b>38</b> <b>Sr</b> 87.62	<b>39</b> <b>Y</b> 88.90585	<b>40</b> <b>Zr</b> 91.224	<b>41</b> <b>Nb</b> 92.90638	<b>42</b> <b>Mo</b> 95.96	<b>43</b> <b>Tc</b> (97.9072)	<b>44</b> <b>Ru</b> 101.07	<b>45</b> <b>Rh</b> 102.90550	<b>46</b> <b>Pd</b> 106.42	<b>47</b> <b>Ag</b> 107.8682	<b>48</b> <b>Cd</b> 112.411	<b>49</b> <b>In</b> 114.818	<b>50</b> <b>Sn</b> 118.710	<b>51</b> <b>Sb</b> 121.760	<b>52</b> <b>Te</b> 127.60	<b>53</b> <b>I</b> 126.90447	<b>54</b> <b>Xe</b> 131.293																												
<b>55</b> <b>Cs</b> 132.9054519	<b>56</b> <b>Ba</b> 137.327	<b>57</b> <b>La</b> 138.90547	<b>72</b> <b>Hf</b> 178.49	<b>73</b> <b>Ta</b> 180.94788	<b>74</b> <b>W</b> 183.84	<b>75</b> <b>Re</b> 186.207	<b>76</b> <b>Os</b> 190.23	<b>77</b> <b>Ir</b> 192.217	<b>78</b> <b>Pt</b> 195.084	<b>79</b> <b>Au</b> 196.966569	<b>80</b> <b>Hg</b> 200.59	<b>81</b> <b>Tl</b> 204.3833*	<b>82</b> <b>Pb</b> 207.2	<b>83</b> <b>Bi</b> 208.98040	<b>84</b> <b>Po</b> (208.9824)	<b>85</b> <b>At</b> (209.9871)	<b>86</b> <b>Rn</b> (222.0176)																												
<b>87</b> <b>Fr</b> (223.0197)	<b>88</b> <b>Ra</b> (226.0254)	<b>89</b> <b>Ac</b> (227.0278)	<b>104</b> <b>Rf</b> (265.1167)	<b>105</b> <b>Db</b> (268.125)	<b>106</b> <b>Sg</b> (271.133)	<b>107</b> <b>Bh</b> (270)	<b>108</b> <b>Hs</b> (277.150)	<b>109</b> <b>Mt</b> (276.151)	<b>110</b> <b>Ds</b> (281.162)	<b>111</b> <b>Rg</b> (280.164)	<b>112</b> <b>Cn</b> (285.174)	<b>113</b> <b>Uut</b> (284.178)	<b>114</b> <b>Fl**</b> (289.187)	<b>115</b> <b>Uup</b> (288.192)	<b>116</b> <b>Lv**</b> (293)	<b>117</b> <b>Uus</b> (294)	<b>118</b> <b>Uuo</b> (294)																												
<table border="1"> <tbody> <tr> <td><b>58</b> <b>Ce</b> 140.116</td> <td><b>59</b> <b>Pr</b> 140.90765</td> <td><b>60</b> <b>Nd</b> 144.242</td> <td><b>61</b> <b>Pm</b> (144.9127)</td> <td><b>62</b> <b>Sm</b> 150.36</td> <td><b>63</b> <b>Eu</b> 151.964</td> <td><b>64</b> <b>Gd</b> 157.25</td> <td><b>65</b> <b>Tb</b> 158.92535</td> <td><b>66</b> <b>Dy</b> 162.500</td> <td><b>67</b> <b>Ho</b> 164.93032</td> <td><b>68</b> <b>Er</b> 167.259</td> <td><b>69</b> <b>Tm</b> 168.93421</td> <td><b>70</b> <b>Yb</b> 173.054</td> <td><b>71</b> <b>Lu</b> 174.9668</td> </tr> <tr> <td><b>90</b> <b>Th</b> 232.03806</td> <td><b>91</b> <b>Pa</b> 231.03588</td> <td><b>92</b> <b>U</b> 238.02891</td> <td><b>93</b> <b>Np</b> (237.0482)</td> <td><b>94</b> <b>Pu</b> (244.0642)</td> <td><b>95</b> <b>Am</b> (243.0614)</td> <td><b>96</b> <b>Cm</b> (247.0704)</td> <td><b>97</b> <b>Bk</b> (247.0703)</td> <td><b>98</b> <b>Cf</b> (251.0796)</td> <td><b>99</b> <b>Es</b> (252.0830)</td> <td><b>100</b> <b>Fm</b> (257.0951)</td> <td><b>101</b> <b>Md</b> (258.0984)</td> <td><b>102</b> <b>No</b> (259.1010)</td> <td><b>103</b> <b>Lr</b> (262.1096)</td> </tr> </tbody> </table>																		<b>58</b> <b>Ce</b> 140.116	<b>59</b> <b>Pr</b> 140.90765	<b>60</b> <b>Nd</b> 144.242	<b>61</b> <b>Pm</b> (144.9127)	<b>62</b> <b>Sm</b> 150.36	<b>63</b> <b>Eu</b> 151.964	<b>64</b> <b>Gd</b> 157.25	<b>65</b> <b>Tb</b> 158.92535	<b>66</b> <b>Dy</b> 162.500	<b>67</b> <b>Ho</b> 164.93032	<b>68</b> <b>Er</b> 167.259	<b>69</b> <b>Tm</b> 168.93421	<b>70</b> <b>Yb</b> 173.054	<b>71</b> <b>Lu</b> 174.9668	<b>90</b> <b>Th</b> 232.03806	<b>91</b> <b>Pa</b> 231.03588	<b>92</b> <b>U</b> 238.02891	<b>93</b> <b>Np</b> (237.0482)	<b>94</b> <b>Pu</b> (244.0642)	<b>95</b> <b>Am</b> (243.0614)	<b>96</b> <b>Cm</b> (247.0704)	<b>97</b> <b>Bk</b> (247.0703)	<b>98</b> <b>Cf</b> (251.0796)	<b>99</b> <b>Es</b> (252.0830)	<b>100</b> <b>Fm</b> (257.0951)	<b>101</b> <b>Md</b> (258.0984)	<b>102</b> <b>No</b> (259.1010)	<b>103</b> <b>Lr</b> (262.1096)
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S.E. Van Bramer 8/29/2012

\* 1995 IUPAC Values from Pure Appl. Chem., Vol 68, No. 12, pp. 2339-2359, 1996. doi: 10.1351/pac199668122339, <http://pac.iupac.org/publications/pac/pdf/1996/pdf/6812x2339.pdf>

\*\*Names for elements 114 and 116 are from Pure Appl. Chem., Vol. 84, No. 7, pp. 1669-1672, 2012. doi: 10.1351/PAC-REC-11-12-03

All other values from: 2009 IUPAC Values from Pure Appl. Chem., Vol. 83, No. 2, pp. 359-396, 2011. doi:10.1351/PAC-REP-10-09-14, <http://pac.iupac.org/publications/pac/pdf/2011/pdf/8302x0359.pdf>

-Elements with one weight have uncertainty in the last digit.

-Elements with the weight in parenthesis, weight is given for the longest lived isotope.